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(54)【発明の名称】 ミネラルウール組成物

(57) [要約]

構成成分が以下の重量パーセンテージ:SIO: 35~60%、好意しくは 39~55%A! : O: 12~27%、好ましくは 16~ 25%CaO 0~85%、好ましくは 3~25%MgO 0~30%、好宏 0~15%Na: O 7%、好ましくは 6~12%Ki O 0~17%、好変しくは 3~12%R2 O (Na: O+K: O) 10~17%、好ましく it 12~17%P: O: 0~ 5 %. 好ましくは 0~ 2%Fe₂ O₃ 0~20%、Bs Os 0~ 8%、好家し <は 0~ 4%TiO₂

%, で示されるところの機能を含み、また、P2 O5 形 態で表現されるリン含布量が、繊維の全質量の8.2% から、特に 0、 5%を超える量から、 5%まで、特に 2 ※未満までの範囲にあり、かつ繊維の表面上に被覆を形 成するように繊維と100℃を超える温度で反応し得る

生理学的媒体中に溶解し得るミネラルウールであって、

ところのリン派化合物も含む。

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CLAIMS

[Claim(s)]

[Claim 1]It is a thermally stable mineral wool which may dissolve in a physiological medium, and constituents are the following weight percentages. : SiO₂35-60%, desirable -- -- 39-55% aluminum _ _ 2O312 to 27% -- desirable -- -- 0 to 35% of CaO 16 to 25%, desirable -- -- 0 to 30% of 3-25% MgO — desirable — — 0-15% Na₂O — 0 to 17%, desirable — — 6-12% K_2 O 0-17% — desirable — -- 3-12% R₂O (Na₂O+K₂O) -- 10 to 17%, It is desirable. Preferably 12 to 17% P₂O₅O to 5% O- 2% $\text{Fe}_2\text{O}_3\text{O}$ =20%, $\text{B}_2\text{O}_3\text{O}$ = 8% -- desirable -- 0 = 4% TiO $\underline{}_2\text{O}$ to 3%, it comes out, and phosphorus content with a P2O5 gestalt including textiles shown, [express and] A mineral wool containing the Lynn system compound which can react to textiles at temperature over 100 ** so that it may be in a range up to less than 2% and covering may be especially formed on the surface of textiles from quantity exceeding especially 0.5% to 5% in 0.2% of total mass of textiles. [Claim 2]Constituents are the following weight percentages. : SiO₂39-55%, desirable -- -- 40-52% aluminum $_{--3 \text{ to } 35\%}$ of CaO 16 to 25% preferably $_2\text{O}_3\text{16}$ to 27%, desirable ---0 to 15% of 10-25% MgO -- desirable -- -- 0-10% Na₂O -- 0 to 15%, desirable -- -- 6-12% K₂O 0-15% -- desirable -- -- 3-12% R₂O (Na₂O+K₂O) -- 10 to 17%, It is desirable. Preferably 12 to 17% P₂O₅O to 5% O- 2% Fe₂O₃O-15%, B₂O₃O - 8% -- desirable -- 0 - 4% TiO -- the mineral wool according to claim 1 which comes out and is characterized by especially MgO being 0 thru/or 2% 0 thru/or 5% when it is $R_2O \le 13.0\%$ 2^0 to 3% including textiles shown.

[Claim 3]Constituents are the following weight percentages.: $SiO_239-55\%$, desirable — — -40-52% aluminum — $_3$ to 35% of CONSTITE OF TOOLS OF TOOLS

[Claim 5]Fe $_2$ O $_3$ (iron whole quantity) content of textiles — 0%<=Fe $_2$ O $_3$ <=5% — desirable — 0% <=Fe $_2$ O $_3$ <=3% — especially — — a mineral wool of Claims 1–4 being 0.5%<=Fe $_2$ O $_3$ <=2.5% given in any 1 paragraph.

[Claim 6]Fe₂O₃ (iron whole quantity) content of textiles -- 5%<=Fe₂O₃<= -- a mineral wool of

Claims 1–4 especially characterized by being 5%<= Fe_2O_3 <=8% 15% given in any 1 paragraph, [Claim 7]A mineral wool of Claims 1–6 given in any 1 paragraph, wherein a presentation of textiles satisfies following related (Na₂O+K₂O) / aluminum₂O₃>=0.5.

[Claim 8]A mineral wool of Claims $\bar{1}$ -7 to which a presentation of textiles is characterized by following relation (Na₂O+K₂O) /aluminum₂O₃>=0.6 and satisfying /aluminum₂O₃>=0.7 especially (Na₂O+K₂O) given in any 1 paragraph.

[Claim 9]lime of textiles, and content of magnesia — 10%<=CaO<=25% — especially — 15% <=CaO<=25% and 0%<=MgO<=5% — desirable — 0%<=MgO<= — a mineral wool of Claims 1-8 especially characterized by being 0%<=MgO<=1% 2% given in any 1 paragraph.

[Claim 10] lime of textiles, and content of magnesia — 5% <= MgO <= 10% and 5% <= CoO <= — a mineral wool of Claims 1-8 characterized by being 5% <= CoO <= 10% preferably 10% given in any 1 paragraph. [Claim 11] A mineral wool of Claims 1-10 having a dissolution rate of 30 ng/cm² even if textiles measure by pH of 4.5 and there are per hour given in any 1 paragraph. [few]

[Claim 12]A mineral wool of Claims 1-11 given in any 1 paragraph, wherein glass equivalent to textiles can fibrose by internal centrifugality.

[Claim 13]A mineral wool of Claims 1–12 given in any 1 paragraph, wherein covering which may be formed on the surface of textiles consists of phosphoric acid alkaline earth metal salt intrinsically. [Claim 14]The mineral wool according to claim 13, wherein phosphoric acid alkaline earth metal salt is calcium phosphate.

[Claim 15]A mineral wool of Claims 1-14 being the compounds which decompose at temperature to which the Lynn system compound which can react to textiles exceeds 100 **, and emit phosphoric acid or a phosphoric anhydride given in any 1 paragraph.

[Claim 16]The mineral wool according to claim 15, wherein the Lynn system compound is chosen from ammonium phosphate, phosphoric acid, and ammonium hide ROGENO phosphate (ammonium hydrogenophosphate).

[Claim 17]It is a manufacturing method of a mineral wool and constituents are the following weight percentages: $\sin O_3 05-60\%$, desirable — -39-55% aluminum $_2 O_3 12$ to 27% — desirable — -0 to 35% of CaO 16 to 25%, desirable — -0 to 30% of 3-25% MgO — desirable — -0-15% Na $_2 O$ — 0 to 17%, desirable — -6-12% K $_2 O$ 0–17% — desirable — -3-12% R $_2 O$ (Na $_2 O$ +K $_2 O$) of 17%, desirable — -12-17% P $_2 O_5 O$ – 5% — desirable — -0 – 2% Fe $_2 O_3 O$ to 20%, B $_2 O_3 O$ – 8% —

desirable --- 0 - 4% TiO __ it coming out, and textiles being formed substantially, and from oxide melt shown, 20 to 3%, [rank second and] How applying the Lynn system compound which can react to textiles in order to form covering on the surface of textiles especially according to a spray or being solution impregnated.

[Claim 18]Use of a mineral wool of Claims 1–16 in a fireproof structure system given in any 1 paragraph.

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JAPANESE	[JP,2003-527287,A]	
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CLAIMS DETAILED DESCRIPTION DRAWINGS

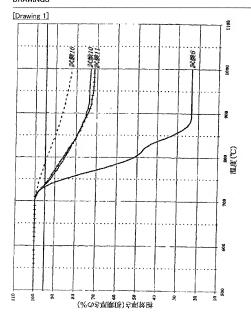
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DRAWINGS



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